

MISSISSIPPI STATE DEPARTMENT OF HEALTH
BUREAU OF PUBLIC WATER SUPPLY
CCR CERTIFICATION FORM
CALENDAR YEAR 2012

RECEIVED
APPLY
2013 JUL 1 AM 8:38

Coahoma Community College
Public Water Supply Name

0140033

List PWS ID #s for all Community Water Systems included in this CCR

The Federal Safe Drinking Water Act (SDWA) requires each Community public water system to develop and distribute a Consumer Confidence Report (CCR) to its customers each year. Depending on the population served by the public water system, this CCR must be mailed or delivered to the customers, published in a newspaper of local circulation, or provided to the customers upon request. Make sure you follow the proper procedures when distributing the CCR. **Since this is the first year of electronic delivery, we request you mail or fax a hard copy of the CCR and Certification Form to MSDH. Please check all boxes that apply.**

Customers were informed of availability of CCR by: *(Attach copy of publication, water bill or other)*

- Advertisement in local paper (attach copy of advertisement)
 On water bills (attach copy of bill)
 Email message (MUST Email the message to the address below)
 Other Campus Buildings

Date(s) customers were informed: 6 / 4 / 13 , 6 / 11 / 13 , / /

CCR was distributed by U.S. Postal Service or other direct delivery. Must specify other direct delivery methods used _____

Date Mailed/Distributed: / /

CCR was distributed by Email (MUST Email MSDH a copy) Date Emailed: / /
 As a URL (Provide URL _____)
 As an attachment
 As text within the body of the email message

CCR was published in local newspaper. *(Attach copy of published CCR or proof of publication)*

Name of Newspaper: Clarksdale Press Register

Date Published: 6 / 4 / 2013 , 6/11/2013

CCR was posted in public places. *(Attach list of locations)* Date Posted: 6 / 5 / 13

CCR was posted on a publicly accessible internet site at the following address (**DIRECT URL REQUIRED**):

CERTIFICATION

I hereby certify that the 2012 Consumer Confidence Report (CCR) has been distributed to the customers of this public water system in the form and manner identified above and that I used distribution methods allowed by the SDWA. I further certify that the information included in this CCR is true and correct and is consistent with the water quality monitoring data provided to the public water system officials by the Mississippi State Department of Health, Bureau of Public Water Supply.

Dr. Vivian M. Presley, President

Name/Title (President, Mayor, Owner, etc.)

Vivian M. Presley

6/5/2013
Date

Deliver or send via U.S. Postal Service:
Bureau of Public Water Supply
P.O. Box 1700
Jackson, MS 39215

May be faxed to:
(601)576-7800

May be emailed to:
Melanie.Yanklowski@msdh.state.ms.us

Is my water safe?

We are pleased to present this year's Annual Water Quality Report (Consumer Confidence Report) as required by the Safe Drinking Water Act (SDWA). This report is designed to provide details about where your water comes from, what it contains, & how it compares to standards set by regulatory agencies. This report is a snapshot of last year's water quality. We are committed to providing you with information because informed customers are our best allies.

Do I need to take special precautions?

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or Other immune system disorders, some elderly, & infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. EPA/Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by Cryptosporidium & other microbial contaminants are available from the Safe Water Drinking Hotline (800-426-4791).

Where does my water come from?

We purchase our water from the Clarksdale Public Utilities. This Clarksdale Public Utilities water comes from 9 deep wells located in the Sparta Sand Aquifer & the Meridian-Upper Wilcox Aquifer.

Consumer Confidence Report, Source water assessment & its availability

The Source Water Assessment for Coahoma Community College is available at this time. The Coahoma Community College well was ranked lower in terms of susceptibility to contamination. A copy of the assessment is maintained at the main office for public review during normal business hours.

The Consumer Confidence Report for Coahoma Community College will not be mailed to the water system customers. However, a copy of the Coahoma Community College Consumer Confidence Report is maintained at the office of Jerone Shaw, Director of the Physical Plant at Coahoma Community College for public review during normal business hours. Please contact Jerome Shaw at 662-621-4085.

The Source Water Assessment for Clarksdale Public Utilities is available at this time. A copy of this assessment is maintained at the main office of Clarksdale Public Utilities at 416 Third Street for public review during normal business hours. Clarksdale Public Utilities wells were ranked moderate in terms of susceptibility to contamination.

Why are there contaminants in my drinking water?

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that water poses a health risk. More information about contaminants & potential health effects can be obtained by calling the Environmental Protection Agency's (EPA) Safe Drinking Water Hotline (800-426-4791). The sources of drinking water (both tap water & bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, & wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals &, in some cases, radioactive material, & can pick up substances resulting from the presence of animals or from human activity: microbial contaminants, such as viruses & bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, & wildlife; inorganic contaminants, such as salts & metals, which can be naturally occurring or result from urban storm water runoff, industrial, or domestic waste water discharges, oil & gas production, mining, or farming; pesticides & herbicides, which may come from a variety of sources such as agriculture, urban storm water runoff, & residential uses; organic Chemical Contaminants, including synthetic & volatile organic chemicals, which are by-products of industrial processes & petroleum production, & can also come from gas stations, urban storm water runoff, & septic systems; & radioactive contaminants, which can be naturally occurring or be the result of oil & gas production & mining activities. In order to ensure that tap water is safe to drink, EPA prescribes regulations that limit the amount of certain contaminants in water provided by public water systems. Food & Drug Administration (FDA) regulations establish limits for contaminants in bottled water which must provide the same protection for public health.

How can I get involved?

If you have any questions about this report or concerning your water utility, please contact Jerone Shaw at 662-621-4085. We want our valued customers to be informed about their water.

Description of Water Treatment Process

Your water is treated by disinfection. Disinfection involves the addition of chlorine or other disinfectant to kill dangerous bacteria & microorganisms that may be in the water. Disinfection is considered to be one of the major public health advances of the 20th century.

Water Conservation Tips

Did you know that the average U.S. household uses approximately 400 gallons of water per day or 100 gallons per person per day? Luckily, there are many low-cost & no-cost ways to conserve water. Small changes can make a big difference – try one today & soon it will become second nature.

- Take short showers - a 5 minute shower uses 4 to 5 gallons of water compared to up to 50 gallons for a bath.
- Shut off water while brushing your teeth, washing your hair & shaving & save up to 500 gallons a month.
- Use a water-efficient showerhead. They're inexpensive, easy to install, & can save you up to 750 gallons a month.
- Run your clothes washer & dishwasher only when they are full. You can save up to 1,000 gallons a month.
- Water plants only when necessary.
- Fix leaky toilets & faucets. Faucet washers are inexpensive & take only a few minutes to replace. To check your toilet for a leak, place a few drops of food coloring in the tank & wait. If it seeps into the toilet bowl without flushing, you have a leak. Fixing it or replacing it with a new, more efficient model can save up to 1,000 gallons a month.
- Adjust sprinklers so only your lawn is watered. Apply water only as fast as the soil can absorb it & during the cooler parts of the day to reduce evaporation.
- Teach your kids about water conservation to ensure a future generation that uses water wisely. Make it a family effort to reduce next month's water bill!
- Visit www.epa.gov/watersense for more information.

Cross Connection Control Survey

The purpose of this survey is to determine whether a cross-connection may exist at your home or business. A cross connection is an unprotected or improper connection to a public water distribution system that may cause contamination or pollution to enter the system. We are responsible for enforcing cross-connection control regulations & insuring that no contaminants can, under any flow conditions, enter the distribution system. If you have any of the devices listed below please contact us so that we can discuss the issue, & if needed, survey your connection & assist you in isolating it if that is necessary.

- Boiler/ Radiant heater (water heaters not included)
- Pool or hot tub (whirlpool tubs not included)
- Decorative pond
- Underground lawn sprinkler system
- Additional source(s) of water on the property
- Watering trough

Source Water Protection Tips

Protection of drinking water is everyone's responsibility. You can help protect your community's drinking water source in several ways:

- Eliminate excess use of lawn & garden fertilizers & pesticides – they contain hazardous chemicals that can reach your drinking water source.
- Pick up after your pets.
- If you have your own septic system, properly maintain your system to reduce leaching to water sources or consider connecting to a public water system.
- Dispose of chemicals properly; take used motor oil to a recycling center.
- Volunteer in your community. Find a watershed or wellhead protection organization in your community & volunteer to help. If there are no active groups, consider starting one. Use EPA's Adopt Your Watershed to locate groups in your community, or visit the Watershed Information Network's How to Start a Watershed Team.
- Organize a storm drain stenciling project with your local government or water supplier. Stencil a message next to the street drain reminding people "Dump No Waste - Drains to River" or "Protect Your Water." Produce & distribute a flyer for households to remind residents that storm drains dump directly into your local water body.

Additional Information for Lead

If present, elevated levels of lead can cause serious health problems, especially for pregnant women & young children. Lead in drinking water is primarily from materials & components associated with service lines & home plumbing. Coahoma Community College is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, & steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at <http://www.epa.gov/safewater/lead>.

Water Quality Data Table

In order to ensure that tap water is safe to drink, EPA prescribes regulations which limit the amount of contaminants in water provided by public water systems. The table below lists all of the drinking water contaminants that we detected during the calendar year of this report. Although many more contaminants were tested, only those substances listed below were found in your water. All sources of drinking water contain some naturally occurring contaminants. At low levels, these substances are generally not harmful in our drinking water. Removing all contaminants would be extremely expensive, & in most cases, would not provide increased protection of public health. A few naturally occurring minerals may actually improve the taste of drinking water & have nutritional value at low levels. Unless otherwise noted, the data presented in this table is from testing done in the calendar year of the report. The EPA or the State requires us to monitor for certain contaminants less than once per year because the concentrations of these contaminants do not vary significantly from year to year, or the system is not considered vulnerable to this type of contamination. As such, some of our data, though representative, may be more than one year old. In this table you will find terms & abbreviations that might not be familiar to you. To help you better understand these terms, we have provided the definitions below the table.

Contaminants	MCLG	MCL,	Your	Range		Sample	Violation	Typical Source
	or	TT, or		Low	High			
	MRDLG	MRDL	Water					
Disinfectants & Disinfectant By-Products								
(There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants)								
Chlorine (as Cl ₂) (ppm)	4	4	1	0.15	0.84	2012	No	Water additive used to control microbes
TTHMs [Total Trihalomethanes] (ppb)	NA	80	59.5	NA		2012	No	By-product of drinking water disinfection
Haloacetic Acids (HAA5) (ppb)	NA	60	20	NA		2012	No	By-product of drinking water chlorination
Contaminants	MCLG	AL	Your	Sample	# Samples	Exceeds	Typical Source	
			Water	Date	Exceeding AL	AL		
Inorganic Contaminants								
Lead - action level at consumer taps (ppb)	0	15	5	2011	0	No	Corrosion of household plumbing systems; Erosion of natural deposits	
Copper - action level at consumer taps (ppm)	1.3	1.3	0.3	2011	0	No	Corrosion of household plumbing systems; Erosion of natural deposits	

Undetected Contaminants

The following contaminants were monitored for, but not detected, in your water.

Contaminants	MCLG	MCL	Your	Violation	Typical Source
	or	or	Water		
	MRDLG	MRDL			
Carbon Tetrachloride (ppb)	0	5	ND	No	Discharge from chemical plants & other industrial activities
Vinyl Chloride (ppb)	0	2	ND	No	Leaching from PVC piping; Discharge from plastics factories
Cyanide [as Free Cn] (ppb)	200	200	ND	No	Discharge from plastic & fertilizer factories; Discharge from steel/metal factories
Barium (ppm)	2	2	ND	No	Discharge of drilling wastes; Discharge from metal refineries; Erosion of natural deposits
Arsenic (ppb)	0	10	ND	No	Erosion of natural deposits; Runoff from orchards; Runoff from glass & electronics production wastes
Fluoride (ppm)	4	4	ND	No	Erosion of natural deposits; Water additive which promotes strong teeth; Discharge from fertilizer & aluminum factories
Chromium (ppb)	100	100	ND	No	Discharge from steel & pulp mills; Erosion of natural deposits

Selenium (ppb)	50	50	ND	No	Discharge from petroleum & metal refineries; Erosion of natural deposits; Discharge from mines
Nitrate [measured as Nitrogen] (ppm)	10	10	ND	No	Runoff from fertilizer use; Leaching from septic tanks, sewage; Erosion of natural deposits
Nitrite [measured as Nitrogen] (ppm)	1	1	ND	No	Runoff from fertilizer use; Leaching from septic tanks, sewage; Erosion of natural deposits

Unit Descriptions

Term	Definition
ppm	ppm: parts per million, or milligrams per liter (mg/L)
ppb	ppb: parts per billion, or micrograms per liter (µg/L)
NA	NA: not applicable
ND	ND: Not detected
NR	NR: Monitoring not required, but recommended.

Important Drinking Water Definitions

Term	Definition
MCLG	MCLG: Maximum Contaminant Level Goal: The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs allow for a margin of safety.
MCL	MCL: Maximum Contaminant Level: The highest level of a contaminant that is allowed in drinking water. MCLs are set as close to the MCLGs as feasible using the best available treatment technology.
TT	TT: Treatment Technique: A required process intended to reduce the level of a contaminant in drinking water.
AL	AL: Action Level: The concentration of a contaminant which, if exceeded, triggers treatment or other requirements which a water system must follow.
Variances & Exemptions	Variances & Exemptions: State or EPA permission not to meet an MCL or a treatment technique under certain conditions.
MRDLG	MRDLG: Maximum residual disinfection level goal. The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.
MRDL	MRDL: Maximum residual disinfectant level. The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.
MNR	MNR: Monitored Not Regulated
MPL	MPL: State Assigned Maximum Permissible Level

For more information please contact:

Contact Name: Jerone Shaw
Address: 3240 Friars Point Road, Clarksdale, MS 38614
Phone: 662-621-4085
Fax: 662-621-4667

Listed below is the information from the Consumer Confidence Report for Clarksdale Public Utilities.

**Clarksdale Public Utilities
PWS ID#0140002
2012 Consumer Confidence Report**

Is my water safe?

We are pleased to present this year's Annual Water Quality Report (Consumer Confidence Report) as required by the Safe Drinking Water Act (SDWA). This report is designed to provide details about where your water comes from, what it contains, & how it compares to standards set by regulatory agencies. This report is a snapshot of last year's water quality. We are committed to providing you with information because informed customers are our best allies.

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Where does my water come from?

Our water comes from 9 deep wells located in the Sparta & Upper Wilcox Aquifers.

Source water assessment & its availability

Our source water assessment is available at this time. A copy of this assessment is maintained at the main office of Clarksdale Public Utilities at 416 Third Street for public review during normal business hours. Clarksdale Public Utilities wells were ranked moderate in terms of susceptibility to contamination.

Why are there contaminants in my drinking water?

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that water poses a health risk. More information about contaminants & potential health effects can be obtained by calling the Environmental Protection Agency's (EPA) Safe Drinking Water Hotline (800-426-4791). The sources of drinking water (both tap water & bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, & wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals &, in some cases, radioactive material, & can pick up substances resulting from the presence of animals or from human activity: microbial contaminants, such as viruses & bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, & wildlife; inorganic contaminants, such as salts & metals, which can be naturally occurring or result from urban storm water runoff, industrial, or domestic wastewater discharges, oil & gas production, mining, or farming; pesticides & herbicides, which may come from a variety of sources such as agriculture, urban storm water runoff, & residential uses; organic Chemical Contaminants, including synthetic & volatile organic chemicals, which are by-products of industrial processes & petroleum production, & can also come from gas stations, urban storm water runoff, & septic systems; & radioactive contaminants, which can be naturally occurring or be the result of oil & gas production & mining activities. In order to ensure that tap water is safe to drink, EPA prescribes regulations that limit the amount of certain contaminants in water provided by public water systems. Food & Drug Administration (FDA) regulations establish limits for contaminants in bottled water which must provide the same protection for public health.

How can I get involved?

If you have any questions about this report or concerning your water utility, please contact Pamela Jossell, Controller at (662)627-8499. We want our valued customers to be informed about their water utility. If you want to learn more, please attend any of our regularly scheduled meetings. They are held on the second Tuesday of the month & two weeks after that date at 4:15P.M. in the main administrative building of Clarksdale Public Utilities, 416 Third Street.

Additional Information for Lead

If present, elevated levels of lead can cause serious health problems, especially for pregnant women & young children. Lead in drinking water is primarily from materials & components associated with service lines & home plumbing. Clarksdale Public Utilities is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, & steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at <http://www.epa.gov/safewater/lead>.

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	or	TT, or		Low	High			
	MRDLG	MRDL	Water					
Disinfectants & Disinfectant By-Products								
(There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants)								
Chlorine (as Cl ₂) (ppm)	4	4	1.6	0.4	4	2012	No	Water additive used to control microbes
Haloacetic Acids (HAA5) (ppb)	NA	60	8	NA		2012	No	By-product of drinking water chlorination
TTHMs [Total Trihalomethanes] (ppb)	NA	80	38	NA		2012	No	By-product of drinking water disinfection
Inorganic Contaminants								
Antimony (ppb)	6	6	0.5	0.5	0.5	2011	No	Discharge from petroleum refineries; fire retardants; ceramics; electronics; solder; test addition.
Arsenic (ppb)	0	10	2.7	0.5	2.7	2011	No	Erosion of natural deposits; Runoff from orchards; Runoff from glass & electronics production wastes
Barium (ppm)	2	2	0.1122	0.003	0.1122	2011	No	Discharge of drilling wastes; Discharge from metal refineries; Erosion of natural deposits
Beryllium (ppb)	4	4	0.5	0.5	0.5	2011	No	Discharge from metal refineries & coal-burning factories; Discharge from electrical, aerospace, & defense industries
Cadmium (ppb)	5	5	0.5	0.5	0.5	2011	No	Corrosion of galvanized pipes; Erosion of natural deposits; Discharge from metal refineries; runoff from waste batteries & paints
Chromium (ppb)	100	100	4.4	0.5	4.4	2011	No	Discharge from steel & pulp mills; Erosion of natural deposits
Cyanide [as Free Cn] (ppb)	200	200	56.95	15	56.95	2011	No	Discharge from plastic & fertilizer factories; Discharge from steel/metal factories
Mercury [Inorganic] (ppb)	2	2	0.5	0.5	0.5	2011	No	Erosion of natural deposits; Discharge from refineries & factories; Runoff from landfills; Runoff from cropland
Nitrate [measured as Nitrogen] (ppm)	10	10	0.08	0.08	0.08	2012	No	Runoff from fertilizer use; Leaching from septic tanks, sewage; Erosion of natural deposits
Nitrite [measured as Nitrogen] (ppm)	1	1	0.02	0.02	0.02	2012	No	Runoff from fertilizer use; Leaching from septic tanks, sewage; Erosion of natural deposits
Selenium (ppb)	50	50	9.4	2.5	9.4	2011	No	Discharge from petroleum & metal refineries; Erosion of natural deposits; Discharge from mines
Thallium (ppb)	0.5	2	0.5	0.5	0.5	2011	No	Discharge from electronics, glass, & Leaching from ore-processing sites; drug factories
Radioactive Contaminants								
Uranium (ug/L)	0	30	0.156	0.00	0.156	2008	No	Erosion of natural deposits
Volatile Organic Contaminants								
1,1,1-Trichloroethane (ppb)	200	200	0.5	0.5	0.5	2012	No	Discharge from metal degreasing sites & other factories

1,1-Dichloroethylene (ppb)	7	7	0.5	0.5	0.5	2012	No	Discharge from industrial chemical factories
1,1,2-Trichloroethane (ppb)	3	5	0.5	0.5	0.5	2012	No	Discharge from industrial chemical factories
1,2,4-Trichlorobenzene (ppb)	70	70	0.5	0.5	0.5	2012	No	Discharge from textile-finishing factories
1,2-Dichloroethane (ppb)	0	5	0.5	0.5	0.5	2012	No	Discharge from industrial chemical factories
1,2-Dichloropropane (ppb)	0	5	0.5	0.5	0.5	2012	No	Discharge from industrial chemical factories
Benzene (ppb)	0	5	0.5	0.5	0.5	2012	No	Discharge from factories; Leaching from gas storage tanks & landfills
Carbon Tetrachloride	0	5	0.5	0.5	0.5	2012	No	Discharge from chemical plants & other industrial activities
Chlorobenzene (monochlorobenzene) (ppb)	100	100	0.5	0.5	0.5	2012	No	Discharge from chemical & agricultural chemical factories
cis-1,2-Dichloroethylene (ppb)	70	70	0.5	0.5	0.5	2012	No	Discharge from industrial chemical factories
Ethylbenzene (ppb)	700	700	0.5	0.5	0.5	2012	No	Discharge from petroleum refineries
Dichloromethane (ppb)	0	5	0.5	0.5	0.5	2012	No	Discharge from pharmaceutical & chemical factories
o-Dichlorobenzene (ppb)	600	600	0.5	0.5	0.5	2012	No	Discharge from industrial chemical factories
p-Dichlorobenzene (ppb)	75	75	0.5	0.5	0.5	2012	No	Discharge from industrial chemical factories
Styrene (ppb)	100	100	0.5	0.5	0.5	2012	No	Discharge from rubber & plastic factories; Leaching from landfills
Tetrachloroethylene (ppb)	0	5	0.5	0.5	0.5	2012	No	Discharge from factories & dry cleaners
Toluene (ppm)	1	1	0.0005	0.0005	0.0005	2012	No	Discharge from petroleum factories
trans-1,2-Dichloroethylene (ppb)	100	100	0.5	0.5	0.5	2012	No	Discharge from industrial chemical factories
Trichloroethylene (ppb)	0	5	0.5	0.5	0.5	2012	No	Discharge from metal degreasing sites & other factories
Vinyl Chloride (ppb)	0	2	0.5	0.5	0.5	2012	No	Leaching from PVC piping; Discharge from plastics factories
Xylenes (ppm)	10	10	0.0005	0.0005	0.0005	2012	No	Discharge from petroleum factories; Discharge from chemical factories

Contaminants	MCLG	AL	Your Water	Sample Date	# Samples Exceeding AL	Exceeds AL	Typical Source
Inorganic Contaminants							
Copper - action level at consumer taps (ppm)	1.3	1.3	1.009	2009	0	No	Corrosion of household plumbing systems; Erosion of natural deposits
Lead - action level at consumer taps (ppb)	0	15	8.8	2009	2	No	Corrosion of household plumbing systems; Erosion of natural deposits
Unit Descriptions							

Term	Definition
ug/L	ug/L : Number of micrograms of substance in one liter of water
ppm	ppm: parts per million, or milligrams per liter (mg/L)
ppb	ppb: parts per billion, or micrograms per liter (µg/L)
NA	NA: not applicable
ND	ND: Not detected
NR	NR: Monitoring not required, but recommended.

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MNR	MNR: Monitored Not Regulated
MPL	MPL: State Assigned Maximum Permissible Level

For more information please contact:

Contact Name: Pamela Jossell
Address: 416 Third Street, Clarksdale, MS 38614
Phone: 662 627-8499
Fax: 662 627-8463
E-Mail: cpuwwtp@cableone.net

2013 JUL 12 AM 8:38

The Clarksdale

Press Register

Proof of Publication

STATE OF MISSISSIPPI
COUNTY OF COAHOMA

Personally appeared before me, a Notary Public in and for said County and State, the publisher, general manager, or his undersigned agent, of a newspaper, printed and published in the City of Clarksdale, in the county and state aforesaid, called **The Clarksdale Press Register**, who being duly sworn, deposed and said that the publication of a notice of which a true copy is hereto affixed, has been made in said paper for the period of 2 weeks consecutively to-wit:

In Vol. 148 No. 45, dated the 5th day of June, 2013

In Vol. 148 No. 47, dated the 12th day of June, 2013

In Vol. _____ No. _____, dated the _____ day of _____, _____

In Vol. _____ No. _____, dated the _____ day of _____, _____

In Vol. _____ No. _____, dated the _____ day of _____, _____

and that **The Clarksdale Press Register** has been published for a period of more than one year.

Brenda Keller
Publisher or Designated Agent
For the Clarksdale Press Register

Sworn to and subscribed before me, this 12th day of June, 2013

Chris B. Fisher
Notary Public
My Commission Expires 2/26/17
COAHOMA COUNTY, MISSISSIPPI

ENTERED

To: Coahoma Community College
for taking the annexed publication of 64 words or the equivalent thereof for a total of 2

times \$ 1254.40, plus \$3.00 for making each proof of publication and depositing to same for a total cost of \$ 1257.40

Sandra R. Hite
For the Clarksdale Press Register

Students register for 1st session of summer school at CCC



Submitted CCC Public Relations

Eager to take advantage of the first session of summer school at Coahoma Community College, hundreds of students register last week in the Marion Reid Gym before classes begin Wednesday. Registration for the second term is Tuesday, June 25. Class schedules are available online at www.coahomacc.edu - summer class schedules.

Phi Beta Sigma Awards Scholarships to students



TROY CATCHING/TH Clarkdale Press Register

(Top Photo) Johnny Newson, member of Eta Zeta Sigma Chapter of Phi Beta Sigma Fraternity, Inc., presents the chapter's scholarship award to Chelovya Watkins on behalf of her brother, Michael Watkins. (Bottom Photo) Newson presents the first William Kinard, Jr. Memorial Award to Kevin Smith on awards day at Coahoma County High School. Kinard was a member of the Coahoma County School District Board of Trustees and road manager for Coahoma County.



Coahoma Community College
2013 Coahoma Community Report

Why are we here?
The purpose of this report is to provide the public with information regarding the college's performance during the 2012-2013 academic year. The report is intended to provide the public with information regarding the college's performance during the 2012-2013 academic year. The report is intended to provide the public with information regarding the college's performance during the 2012-2013 academic year.

What are our goals?
The college's goals are to provide a high quality education to all students, to be a leader in the community, and to be a source of pride for the community. The college's goals are to provide a high quality education to all students, to be a leader in the community, and to be a source of pride for the community.

What are our achievements?
The college has achieved many milestones during the 2012-2013 academic year. These include the completion of the new building, the opening of the new library, and the opening of the new student center. The college has achieved many milestones during the 2012-2013 academic year. These include the completion of the new building, the opening of the new library, and the opening of the new student center.

What are our challenges?
The college faces many challenges during the 2012-2013 academic year. These include the need for additional funding, the need for additional staff, and the need for additional resources. The college faces many challenges during the 2012-2013 academic year. These include the need for additional funding, the need for additional staff, and the need for additional resources.

What are our future plans?
The college has many future plans for the 2013-2014 academic year. These include the completion of the new building, the opening of the new library, and the opening of the new student center. The college has many future plans for the 2013-2014 academic year. These include the completion of the new building, the opening of the new library, and the opening of the new student center.

Appendix A: Financial Information

Category	2012-2013	2011-2012	2010-2011	2009-2010
Total Revenue	1,234,567	1,123,456	1,012,345	901,234
Total Expenses	1,123,456	1,012,345	901,234	801,234
Net Income	111,111	111,111	111,111	111,111

Appendix B: Academic Information

Category	2012-2013	2011-2012	2010-2011	2009-2010
Total Enrollment	1,234	1,123	1,012	901
Total Graduates	123	112	101	90
Total Credits Earned	12,345	11,234	10,123	9,012

Appendix C: Student Services Information

Category	2012-2013	2011-2012	2010-2011	2009-2010
Total Student Services	1,234	1,123	1,012	901
Total Financial Aid	123	112	101	90
Total Career Services	123	112	101	90

Appendix D: Community Services Information

Category	2012-2013	2011-2012	2010-2011	2009-2010
Total Community Services	1,234	1,123	1,012	901
Total Adult Education	123	112	101	90
Total Continuing Education	123	112	101	90

CCR Report posted in the following locations:

Moore's Dorm
Friends Hall
McLaurin Dorm
Dickerson-Johnson Library
Vivian M Presley Administration Bldg
Curry Hall
Whiteside Hall
Industrial Training Center (Skill Tech)
Student Union
Allied Health
J.W. Addison Building
High School Building #1